

(3*R,4*R**)-1-(4-Chlorophenyl)-4-[2-hydroxy-3-(morpholinomethyl)phenyl]-3-phenoxyazetid-2-one**Mehmet Akkurt,^{a*} Aliasghar Jarrahpour,^b Hashem Sharghi,^b Seid Ali Torabi Badrabad^b and Orhan Büyükgüngör^c^aDepartment of Physics, Faculty of Sciences, Erciyes University, 38039 Kayseri, Turkey, ^bDepartment of Chemistry, College of Sciences, Shiraz University, 71454 Shiraz, Iran, and ^cDepartment of Physics, Faculty of Arts and Sciences, Ondokuz Mayıs University, 55139 Samsun, Turkey
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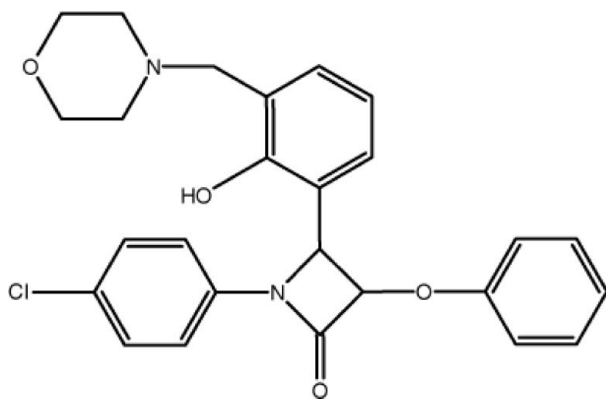
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.008$ Å; R factor = 0.087; wR factor = 0.130; data-to-parameter ratio = 15.6.

The β -lactam ring of the title compound, $\text{C}_{26}\text{H}_{25}\text{ClN}_2\text{O}_4$, is nearly planar (r.m.s. deviation = 0.025 Å) and the morpholine ring adopts a chair conformation. The mean plane of the β -lactam ring makes dihedral angles of 21.6 (4), 84.4 (4) and 33.7 (4)° with the two benzene rings and the phenyl ring, respectively. The conformation of the title compound is stabilized by intramolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{O}-\text{H}\cdots\text{N}$ interactions. The crystal structure features $\text{C}-\text{H}\cdots\pi$ and aromatic $\pi-\pi$ stacking interactions [centroid-centroid distances = 3.684 (4) and 3.883 (4) Å].

Related literature

For a related structure, see: Akkurt *et al.* (2011). For puckering parameters, see: Cremer & Pople (1975).

**Experimental***Crystal data* $\text{C}_{26}\text{H}_{25}\text{ClN}_2\text{O}_4$
 $M_r = 464.93$
Monoclinic, $C2/c$
 $a = 29.6418$ (18) Å
 $b = 6.7166$ (3) Å
 $c = 28.5708$ (15) Å
 $\beta = 123.043$ (4)°
 $V = 4768.2$ (5) Å³
 $Z = 8$
Mo $K\alpha$ radiation
 $\mu = 0.20$ mm⁻¹
 $T = 296$ K
 $0.53 \times 0.23 \times 0.04$ mm*Data collection*Stoe IPDS 2 diffractometer
Absorption correction: integration (*X-RED32*; Stoe & Cie, 2002)
 $T_{\min} = 0.904$, $T_{\max} = 0.992$
16824 measured reflections
4707 independent reflections
1712 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.159$ *Refinement* $R[F^2 > 2\sigma(F^2)] = 0.087$
 $wR(F^2) = 0.130$
 $S = 1.00$
4707 reflections
302 parameters
1 restraint
H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.15$ e Å⁻³
 $\Delta\rho_{\min} = -0.15$ e Å⁻³**Table 1**

Hydrogen-bond geometry (Å, °).

Cg5 is a centroid of the C16–C21 benzene ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O3—H3A \cdots N2	0.85 (6)	1.87 (6)	2.642 (8)	150 (5)
C2—H2 \cdots O2	0.93	2.50	3.304 (6)	144
C15—H15 \cdots O2	0.93	2.54	3.148 (7)	123
C20—H20 \cdots Cg5 ⁱ	0.93	2.88	3.596 (5)	134

Symmetry code: (i) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5782).

References

- Akkurt, M., Dağdemir, Y., Jarrahpour, A., Rostami, M. & Büyükgüngör, O. (2011). *Acta Cryst.* **E67**, o326–o327.
Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Stoe & Cie (2002). *X-AREA* and *X-RED32*. Stoe & Cie, Darmstadt, Germany.

supplementary materials

Acta Cryst. (2011). E67, o325 [doi:10.1107/S1600536811000675]

(3*R,4*R**)-1-(4-Chlorophenyl)-4-[2-hydroxy-3-(morpholinomethyl)phenyl]-3-phenoxyazetid-2-one**

M. Akkurt, A. Jarrahpour, H. Sharghi, S. A. T. Badrabad and O. Büyükgüngör

Comment

As part of our ongoing structural studies of β -lactams (Akkurt *et al.*, 2011), we now report the structure of the title compound, (I).

In the title compound, (I), (Fig. 1), the β -lactam ring (N1/C7–C9) is almost planar, with long C—C distances [C7–C9 = 1.518 (7) Å and C7–C8 = 1.553 (6) Å]. The morpholine ring (N2/O4/C23–C26) adopts a chair conformation, with the puckering parameters of $Q_T = 0.594$ (8) Å, $\theta = 177.1$ (8)° and $\varphi = 305$ (17)° (Cremer & Pople, 1975). The dihedral angles between the mean planes of the rings in (I) are given in Table 2.

The molecular conformation of (I) is stabilized by intra-molecular C—H \cdots O and O—H \cdots N interactions (Table 1). The crystal structure is stabilized by C—H $\cdots\pi$ interactions (Table 1) and two π - π stacking interactions [Cg1 \cdots Cg5(*x*, *y*, *z*) = 3.684 (4) Å and Cg4 \cdots Cg4(1 - *x*, -*y*, -*z*) = 3.883 (4) Å, where Cg1, Cg4 and Cg5 are the centroids of the N1/C7–C9 β -lactam, the C10–C15 and C16–C21 benzene rings, respectively]. Fig. 2 shows the packing of (I), down the *b* axis.

Experimental

To a solution of (*E*)-2-((4-chlorophenylimino)methyl)-6-(morpholinomethyl)phenol (1.0 mmol) and triethylamine (2.6 mmol) in dry CH₂Cl₂ was slowly added phenoxyacetyl chloride (1.3 mmol) in dry CH₂Cl₂ (10 ml) at 195 K. The reaction mixture was then allowed to warm to room temperature, stirred over night and then it was washed with saturated sodium bicarbonate solution (20 ml), brine (20 ml), dried (Na₂SO₄). The solvent was evaporated under reduced pressure to give the crude product which was then purified by column chromatography over silica gel. Colourless needles were recrystallised from ethyl acetate (yield 65%). [mp: 463–465 K]. IR (KBr, cm⁻¹): 1758.5 (CO β -Lactam), 3311–3497 (OH). ¹H-NMR (250 MHz, CDCl₃) δ (p.p.m): 3.43 (CH₂—N, t, 4H, *J* = 13.5 Hz), 3.70 (CH₂, s, 2H), 3.88 (CH₂—O, t, 4H, *J* = 13.9 Hz), 5.59 (H-8, d, 1H, *J* = 1.6 Hz), 5.91 (H-7, d, 1H, *J* = 1.6 Hz), 6.67–7.35 (ArH, m, 12H); ¹³C-NMR (62.9 MHz, CDCl₃) δ (p.p.m): 52.6 (CH₂—N), 55.9 (C-22), 50.2 (O—CH₂), 66.5 (C-8), 80.9 (C-7), 115.2–157.0 (aromatic carbons), 163.1 (CO β -Lactam). Analysis calculated for C₂₆H₂₅ClN₂O₄: C 67.17, H 5.42, N 6.03%. found: C 67.05, H 5.47, N 6.08%.

Refinement

The position of the bridging hydroxyl hydrogen atom was found in a difference Fourier map, and refined freely by the soft-constraint method of the hydrogen-position of 0.83 (2) Å. The other H atoms were placed at calculated positions and were treated as riding on their parent atoms with O—H = 0.82 Å, C—H = 0.93 (aromatic), 0.96 (methyl), 0.97 Å (methyline) and 0.98 Å (methine), and with $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl and $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic, methine and methyline.

Figures

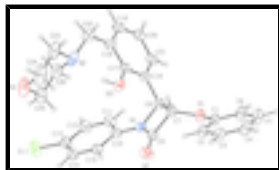


Fig. 1. The molecular structure of (I) with displacement ellipsoids for non-H atoms drawn at the 20% probability level.



Fig. 2. The packing of (I), viewing down the *b* axis. All H atoms are omitted for clarity.

(3*R**,4*R**)-1-(4-Chlorophenyl)-4-[2-hydroxy-3-(morpholinomethyl)phenyl]-3-phenoxyazetid-2-one

Crystal data

$C_{26}H_{25}ClN_2O_4$

$M_r = 464.93$

Monoclinic, *C2/c*

Hall symbol: -C 2yc

$a = 29.6418 (18) \text{ \AA}$

$b = 6.7166 (3) \text{ \AA}$

$c = 28.5708 (15) \text{ \AA}$

$\beta = 123.043 (4)^\circ$

$V = 4768.2 (5) \text{ \AA}^3$

$Z = 8$

$F(000) = 1952$

$D_x = 1.295 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 9239 reflections

$\theta = 1.4\text{--}26.6^\circ$

$\mu = 0.20 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Needle, colourless

$0.53 \times 0.23 \times 0.04 \text{ mm}$

Data collection

Stoe IPDS 2
diffractometer

4707 independent reflections

Radiation source: sealed X-ray tube, 12 x 0.4 mm
long-fine focus

1712 reflections with $I > 2\sigma(I)$

plane graphite

$R_{\text{int}} = 0.159$

Detector resolution: 6.67 pixels mm^{-1}

$\theta_{\text{max}} = 26.2^\circ$, $\theta_{\text{min}} = 1.6^\circ$

ω scans

$h = -36 \rightarrow 36$

Absorption correction: integration
(*X-RED32*; Stoe & Cie, 2002)

$k = -8 \rightarrow 7$

$T_{\text{min}} = 0.904$, $T_{\text{max}} = 0.992$

$l = -34 \rightarrow 32$

16824 measured reflections

Refinement

Refinement on F^2

Primary atom site location: structure-invariant direct methods

Least-squares matrix: full

Secondary atom site location: difference Fourier map

$$R[F^2 > 2\sigma(F^2)] = 0.087$$

$$wR(F^2) = 0.130$$

$$S = 1.00$$

4707 reflections

302 parameters

1 restraint

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0198P)^2]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.15 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.15 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.55205 (8)	-0.0943 (3)	-0.11446 (7)	0.1225 (9)
O1	0.58355 (13)	0.2225 (4)	0.21000 (14)	0.0705 (15)
O2	0.53938 (15)	0.4640 (5)	0.09015 (15)	0.0921 (16)
O3	0.67681 (14)	0.3208 (4)	0.13843 (16)	0.0711 (15)
O4	0.7152 (2)	0.5197 (7)	0.0038 (2)	0.133 (3)
N1	0.57506 (14)	0.1426 (5)	0.09621 (17)	0.0565 (16)
N2	0.74627 (19)	0.3438 (6)	0.1079 (2)	0.080 (2)
C1	0.58006 (18)	0.3860 (7)	0.2379 (2)	0.064 (2)
C2	0.57298 (19)	0.5775 (7)	0.2178 (2)	0.073 (2)
C3	0.5692 (2)	0.7320 (7)	0.2476 (3)	0.085 (3)
C4	0.5719 (2)	0.6941 (8)	0.2963 (3)	0.097 (3)
C5	0.5791 (2)	0.5024 (9)	0.3163 (3)	0.103 (3)
C6	0.5825 (2)	0.3474 (7)	0.2858 (2)	0.082 (3)
C7	0.60357 (19)	0.2554 (6)	0.1756 (2)	0.059 (2)
C8	0.61155 (18)	0.0579 (6)	0.15272 (19)	0.0549 (17)
C9	0.5657 (2)	0.3163 (7)	0.1150 (2)	0.064 (2)
C10	0.56823 (18)	0.0872 (7)	0.0456 (2)	0.0600 (17)
C11	0.58087 (19)	-0.1071 (6)	0.0384 (2)	0.071 (2)
C12	0.5758 (2)	-0.1594 (7)	-0.0105 (2)	0.077 (2)
C13	0.5587 (2)	-0.0228 (9)	-0.0526 (2)	0.080 (2)
C14	0.5446 (2)	0.1665 (8)	-0.0465 (2)	0.077 (2)
C15	0.54974 (18)	0.2211 (7)	0.0022 (2)	0.067 (2)
C16	0.66644 (18)	0.0030 (6)	0.16733 (18)	0.0544 (16)

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C17	0.69762 (19)	0.1335 (6)	0.15820 (19)	0.0580 (17)
C18	0.7471 (2)	0.0758 (8)	0.1678 (2)	0.067 (2)
C19	0.7664 (2)	-0.1138 (8)	0.1899 (2)	0.078 (2)
C20	0.7373 (2)	-0.2423 (8)	0.2011 (2)	0.077 (2)
C21	0.6879 (2)	-0.1842 (6)	0.18915 (19)	0.0653 (19)
C22	0.7803 (2)	0.2180 (8)	0.1572 (3)	0.093 (3)
C23	0.7780 (3)	0.5021 (9)	0.1032 (3)	0.117 (4)
C24	0.7411 (4)	0.6337 (10)	0.0543 (4)	0.145 (5)
C25	0.6841 (2)	0.3629 (9)	0.0067 (3)	0.111 (3)
C26	0.7192 (2)	0.2287 (7)	0.0555 (2)	0.088 (3)
H2	0.57070	0.60280	0.18450	0.0890*
H3	0.56490	0.86200	0.23460	0.1020*
H3A	0.6940 (18)	0.370 (7)	0.125 (2)	0.093 (19)*
H4	0.56880	0.79830	0.31580	0.1160*
H5	0.58170	0.47660	0.34970	0.1240*
H6	0.58640	0.21700	0.29830	0.0990*
H7	0.63640	0.33630	0.19450	0.0700*
H8	0.59450	-0.05380	0.15940	0.0650*
H11	0.59270	-0.20010	0.06690	0.0840*
H12	0.58390	-0.28860	-0.01530	0.0920*
H14	0.53160	0.25730	-0.07570	0.0920*
H15	0.54070	0.34970	0.00620	0.0810*
H19	0.79970	-0.15400	0.19720	0.0940*
H20	0.75100	-0.36680	0.21670	0.0920*
H21	0.66790	-0.27270	0.19580	0.0780*
H22A	0.80150	0.30230	0.18970	0.1110*
H22B	0.80490	0.14230	0.15170	0.1110*
H23A	0.80490	0.44350	0.09800	0.1410*
H23B	0.79640	0.58040	0.13720	0.1410*
H24A	0.71410	0.69200	0.05950	0.1730*
H24B	0.76160	0.74110	0.05180	0.1730*
H25A	0.66690	0.28600	-0.02760	0.1340*
H25B	0.65610	0.41960	0.01050	0.1340*
H26A	0.69750	0.12460	0.05720	0.1060*
H26B	0.74600	0.16630	0.05070	0.1060*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.1695 (18)	0.1205 (13)	0.0933 (13)	0.0114 (11)	0.0819 (13)	-0.0060 (10)
O1	0.100 (3)	0.0508 (19)	0.086 (3)	-0.0019 (17)	0.067 (2)	0.0026 (17)
O2	0.115 (3)	0.070 (2)	0.091 (3)	0.038 (2)	0.056 (2)	0.023 (2)
O3	0.089 (3)	0.0445 (18)	0.106 (3)	-0.0007 (17)	0.070 (2)	0.0045 (17)
O4	0.208 (5)	0.098 (3)	0.168 (5)	-0.024 (3)	0.150 (5)	0.006 (3)
N1	0.061 (3)	0.050 (2)	0.059 (3)	0.0081 (19)	0.033 (2)	0.006 (2)
N2	0.103 (4)	0.060 (3)	0.113 (4)	-0.023 (3)	0.083 (3)	-0.021 (3)
C1	0.067 (4)	0.059 (3)	0.084 (4)	-0.003 (2)	0.052 (3)	0.001 (3)
C2	0.088 (4)	0.060 (3)	0.092 (4)	0.001 (3)	0.062 (4)	0.007 (3)

C3	0.105 (5)	0.058 (3)	0.124 (5)	0.003 (3)	0.083 (4)	0.003 (3)
C4	0.118 (5)	0.076 (4)	0.140 (6)	-0.001 (4)	0.098 (5)	-0.017 (4)
C5	0.145 (6)	0.089 (4)	0.126 (6)	-0.001 (4)	0.107 (5)	-0.003 (4)
C6	0.117 (5)	0.062 (3)	0.107 (5)	-0.003 (3)	0.086 (4)	0.003 (3)
C7	0.074 (4)	0.051 (3)	0.067 (4)	-0.001 (2)	0.049 (3)	0.003 (2)
C8	0.060 (3)	0.044 (3)	0.064 (3)	-0.005 (2)	0.036 (3)	0.002 (2)
C9	0.066 (4)	0.058 (3)	0.072 (4)	0.004 (3)	0.041 (3)	0.010 (3)
C10	0.056 (3)	0.057 (3)	0.062 (3)	-0.002 (2)	0.029 (3)	0.005 (3)
C11	0.083 (4)	0.048 (3)	0.067 (4)	-0.001 (3)	0.032 (3)	0.000 (3)
C12	0.090 (4)	0.063 (3)	0.072 (4)	0.008 (3)	0.040 (4)	-0.009 (3)
C13	0.091 (4)	0.089 (4)	0.060 (4)	0.002 (3)	0.042 (4)	-0.004 (3)
C14	0.084 (4)	0.077 (4)	0.069 (4)	0.011 (3)	0.041 (3)	0.015 (3)
C15	0.073 (4)	0.064 (3)	0.070 (4)	0.014 (3)	0.042 (3)	0.005 (3)
C16	0.063 (3)	0.041 (2)	0.060 (3)	0.001 (2)	0.034 (3)	-0.001 (2)
C17	0.063 (3)	0.046 (3)	0.065 (3)	0.007 (2)	0.035 (3)	-0.002 (2)
C18	0.057 (4)	0.073 (3)	0.074 (4)	-0.001 (3)	0.037 (3)	-0.009 (3)
C19	0.067 (4)	0.081 (4)	0.069 (4)	0.019 (3)	0.026 (3)	-0.009 (3)
C20	0.079 (4)	0.063 (3)	0.069 (4)	0.017 (3)	0.027 (3)	0.011 (3)
C21	0.066 (4)	0.053 (3)	0.065 (3)	0.007 (3)	0.028 (3)	0.008 (2)
C22	0.084 (5)	0.093 (4)	0.118 (5)	-0.009 (4)	0.066 (4)	-0.010 (4)
C23	0.165 (7)	0.083 (4)	0.173 (7)	-0.048 (5)	0.137 (6)	-0.041 (5)
C24	0.250 (10)	0.069 (4)	0.208 (10)	-0.038 (5)	0.185 (9)	-0.017 (5)
C25	0.139 (6)	0.100 (5)	0.124 (6)	-0.009 (4)	0.091 (5)	0.015 (4)
C26	0.108 (5)	0.068 (3)	0.115 (5)	-0.014 (3)	0.078 (4)	-0.012 (4)

Geometric parameters (Å, °)

C11—C13	1.736 (6)	C17—C18	1.392 (9)
O1—C1	1.394 (6)	C18—C22	1.516 (9)
O1—C7	1.415 (7)	C18—C19	1.398 (8)
O2—C9	1.222 (6)	C19—C20	1.376 (9)
O3—C17	1.379 (5)	C20—C21	1.367 (9)
O4—C24	1.432 (10)	C23—C24	1.504 (11)
O4—C25	1.432 (9)	C25—C26	1.502 (8)
O3—H3A	0.85 (6)	C2—H2	0.9300
N1—C9	1.374 (6)	C3—H3	0.9300
N1—C10	1.397 (7)	C4—H4	0.9300
N1—C8	1.482 (6)	C5—H5	0.9300
N2—C22	1.472 (8)	C6—H6	0.9300
N2—C23	1.473 (10)	C7—H7	0.9800
N2—C26	1.474 (7)	C8—H8	0.9800
C1—C2	1.377 (7)	C11—H11	0.9300
C1—C6	1.356 (7)	C12—H12	0.9300
C2—C3	1.385 (8)	C14—H14	0.9300
C3—C4	1.373 (10)	C15—H15	0.9300
C4—C5	1.377 (8)	C19—H19	0.9300
C5—C6	1.396 (8)	C20—H20	0.9300
C7—C9	1.518 (7)	C21—H21	0.9300
C7—C8	1.553 (6)	C22—H22A	0.9700

supplementary materials

C8—C16	1.489 (8)	C22—H22B	0.9700
C10—C15	1.380 (7)	C23—H23A	0.9700
C10—C11	1.404 (7)	C23—H23B	0.9700
C11—C12	1.367 (7)	C24—H24A	0.9700
C12—C13	1.371 (7)	C24—H24B	0.9700
C13—C14	1.379 (8)	C25—H25A	0.9700
C14—C15	1.365 (7)	C25—H25B	0.9700
C16—C17	1.398 (8)	C26—H26A	0.9700
C16—C21	1.394 (6)	C26—H26B	0.9700
C1—O1—C7	117.7 (4)	C3—C2—H2	120.00
C24—O4—C25	110.1 (6)	C2—C3—H3	120.00
C17—O3—H3A	107 (3)	C4—C3—H3	120.00
C8—N1—C9	94.8 (4)	C3—C4—H4	120.00
C9—N1—C10	133.2 (4)	C5—C4—H4	120.00
C8—N1—C10	129.3 (4)	C4—C5—H5	120.00
C22—N2—C23	111.3 (5)	C6—C5—H5	120.00
C23—N2—C26	107.9 (5)	C1—C6—H6	120.00
C22—N2—C26	112.4 (4)	C5—C6—H6	120.00
O1—C1—C6	116.6 (4)	O1—C7—H7	112.00
C2—C1—C6	120.8 (5)	C8—C7—H7	112.00
O1—C1—C2	122.6 (5)	C9—C7—H7	112.00
C1—C2—C3	119.2 (5)	N1—C8—H8	111.00
C2—C3—C4	120.3 (5)	C7—C8—H8	111.00
C3—C4—C5	120.3 (6)	C16—C8—H8	111.00
C4—C5—C6	119.1 (6)	C10—C11—H11	120.00
C1—C6—C5	120.4 (5)	C12—C11—H11	120.00
O1—C7—C8	112.1 (4)	C11—C12—H12	120.00
O1—C7—C9	120.3 (5)	C13—C12—H12	120.00
C8—C7—C9	86.4 (4)	C13—C14—H14	120.00
N1—C8—C7	86.7 (3)	C15—C14—H14	120.00
N1—C8—C16	115.9 (4)	C10—C15—H15	120.00
C7—C8—C16	119.2 (4)	C14—C15—H15	120.00
O2—C9—C7	136.2 (5)	C18—C19—H19	119.00
N1—C9—C7	92.0 (4)	C20—C19—H19	119.00
O2—C9—N1	131.6 (5)	C19—C20—H20	121.00
N1—C10—C11	120.0 (4)	C21—C20—H20	120.00
N1—C10—C15	121.3 (4)	C16—C21—H21	119.00
C11—C10—C15	118.8 (5)	C20—C21—H21	119.00
C10—C11—C12	120.0 (4)	N2—C22—H22A	109.00
C11—C12—C13	120.4 (5)	N2—C22—H22B	109.00
C11—C13—C14	120.7 (4)	C18—C22—H22A	109.00
C12—C13—C14	119.9 (5)	C18—C22—H22B	109.00
C11—C13—C12	119.4 (5)	H22A—C22—H22B	108.00
C13—C14—C15	120.2 (5)	N2—C23—H23A	110.00
C10—C15—C14	120.6 (5)	N2—C23—H23B	110.00
C8—C16—C17	121.7 (4)	C24—C23—H23A	110.00
C8—C16—C21	120.8 (5)	C24—C23—H23B	110.00
C17—C16—C21	117.5 (5)	H23A—C23—H23B	108.00
O3—C17—C18	121.5 (5)	O4—C24—H24A	110.00

C16—C17—C18	121.4 (4)	O4—C24—H24B	110.00
O3—C17—C16	117.1 (5)	C23—C24—H24A	110.00
C17—C18—C19	118.1 (5)	C23—C24—H24B	110.00
C19—C18—C22	120.9 (6)	H24A—C24—H24B	108.00
C17—C18—C22	121.0 (5)	O4—C25—H25A	109.00
C18—C19—C20	121.5 (6)	O4—C25—H25B	110.00
C19—C20—C21	119.0 (5)	C26—C25—H25A	110.00
C16—C21—C20	122.3 (5)	C26—C25—H25B	109.00
N2—C22—C18	111.9 (6)	H25A—C25—H25B	108.00
N2—C23—C24	109.4 (7)	N2—C26—H26A	110.00
O4—C24—C23	109.9 (6)	N2—C26—H26B	110.00
O4—C25—C26	110.7 (6)	C25—C26—H26A	110.00
N2—C26—C25	110.2 (4)	C25—C26—H26B	109.00
C1—C2—H2	120.00	H26A—C26—H26B	108.00
C7—O1—C1—C2	27.8 (8)	O1—C7—C9—N1	-115.5 (4)
C7—O1—C1—C6	-153.9 (5)	C9—C7—C8—C16	119.7 (5)
C1—O1—C7—C8	173.3 (4)	C8—C7—C9—N1	-1.9 (4)
C1—O1—C7—C9	-87.5 (5)	C9—C7—C8—N1	1.7 (4)
C24—O4—C25—C26	-58.8 (8)	C7—C8—C16—C21	128.8 (5)
C25—O4—C24—C23	60.3 (11)	N1—C8—C16—C21	-129.8 (4)
C10—N1—C8—C7	161.0 (5)	N1—C8—C16—C17	48.2 (6)
C9—N1—C8—C16	-122.9 (4)	C7—C8—C16—C17	-53.2 (6)
C9—N1—C10—C11	-178.9 (6)	N1—C10—C15—C14	178.3 (6)
C10—N1—C9—C7	-159.8 (6)	N1—C10—C11—C12	-177.9 (6)
C8—N1—C10—C15	-154.4 (5)	C15—C10—C11—C12	1.3 (9)
C10—N1—C8—C16	40.0 (7)	C11—C10—C15—C14	-1.0 (9)
C8—N1—C9—C7	2.0 (5)	C10—C11—C12—C13	0.4 (9)
C8—N1—C9—O2	177.5 (7)	C11—C12—C13—C11	-179.4 (5)
C10—N1—C9—O2	15.7 (12)	C11—C12—C13—C14	-2.4 (10)
C9—N1—C10—C15	1.9 (10)	C12—C13—C14—C15	2.8 (10)
C8—N1—C10—C11	24.8 (9)	C11—C13—C14—C15	179.7 (5)
C9—N1—C8—C7	-1.9 (5)	C13—C14—C15—C10	-1.1 (9)
C23—N2—C26—C25	-58.2 (7)	C17—C16—C21—C20	-0.5 (7)
C22—N2—C23—C24	-176.7 (6)	C21—C16—C17—C18	3.2 (7)
C26—N2—C22—C18	-68.0 (7)	C8—C16—C17—O3	4.2 (6)
C26—N2—C23—C24	59.6 (8)	C21—C16—C17—O3	-177.8 (4)
C22—N2—C26—C25	178.7 (6)	C8—C16—C21—C20	177.5 (4)
C23—N2—C22—C18	171.0 (5)	C8—C16—C17—C18	-174.9 (4)
O1—C1—C6—C5	-179.8 (6)	C16—C17—C18—C19	-3.5 (7)
C6—C1—C2—C3	1.0 (9)	O3—C17—C18—C22	0.4 (7)
O1—C1—C2—C3	179.2 (6)	O3—C17—C18—C19	177.6 (4)
C2—C1—C6—C5	-1.5 (9)	C16—C17—C18—C22	179.4 (5)
C1—C2—C3—C4	-0.7 (10)	C19—C18—C22—N2	146.3 (5)
C2—C3—C4—C5	0.9 (11)	C17—C18—C22—N2	-36.7 (7)
C3—C4—C5—C6	-1.3 (11)	C22—C18—C19—C20	178.3 (5)
C4—C5—C6—C1	1.6 (10)	C17—C18—C19—C20	1.1 (7)
O1—C7—C9—O2	69.3 (9)	C18—C19—C20—C21	1.4 (7)
O1—C7—C8—N1	123.1 (4)	C19—C20—C21—C16	-1.7 (7)
C8—C7—C9—O2	-177.1 (8)	N2—C23—C24—O4	-61.4 (11)

supplementary materials

O1—C7—C8—C16

-118.9 (5)

O4—C25—C26—N2

58.4 (7)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg5 is a centroid of the C16–C21 benzene ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O3—H3A \cdots N2	0.85 (6)	1.87 (6)	2.642 (8)	150 (5)
C2—H2 \cdots O2	0.93	2.50	3.304 (6)	144
C15—H15 \cdots O2	0.93	2.54	3.148 (7)	123
C20—H20 \cdots Cg5 ⁱ	0.93	2.88	3.596 (5)	134

Symmetry codes: (i) $-x+3/2, y-1/2, -z+1/2$.

Table 2

The dihedral angles between the mean planes of the rings in (I) ($^\circ$)

	Ring-2	Ring-3	Ring-4
Ring-1	33.7 (4)	21.6 (4)	84.4 (4)
Ring-2		13.0 (3)	63.0 (3)
Ring-3			67.6 (3)

Ring-1 : N1/C7–C9 β -lactam ring, Ring-2 : C1–C6 phenyl ring, Ring-3 : C10–C15 benzene ring, Ring-4 : C16–C21 benzene ring.

Fig. 1

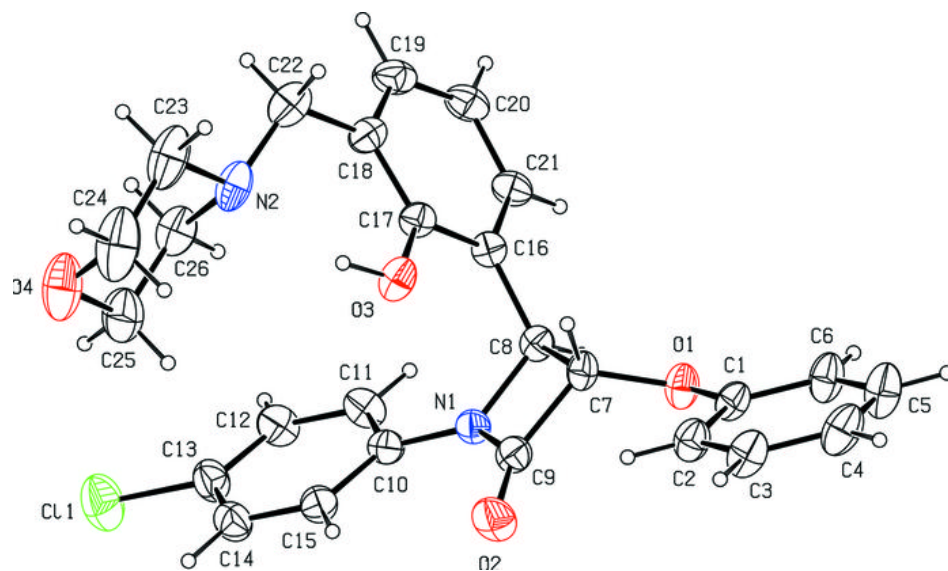


Fig. 2

